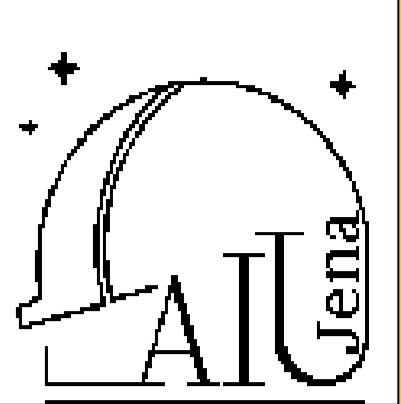


# Semiconductor- & Metallic-Behavior of Iron(II) Sulfides: A Multi-Wavelengths Study



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## Abstract

- Iron(II) sulfides: **Sulfur reservoirs in interplanetary dust** (e.g. in comets, meteorites, IDPs)
- Prior spectral studies: Covers either the UV-VIS to NIR, or the FIR spectral range
- Our study: Reflectivity data from **VUV to FIR** for the determination of optical constants
- Five iron(II) sulfides with different Fe/S ratio: From **4C-pyrrhotite to troilite**
- Binary spectroscopic behavior** between stoichiometric(-like) & iron depleted iron(II) sulfides
- Optical constants** depend on Fe/S ratio & **IR reflectivity decreases with higher iron amount**

## Previous Optical Studies

- [1] W. G. Egan, T. Hilgeman, The interstellar medium: UV complex index of refraction of several candidate materials, *Astronomical Journal* 80, 1975, pp. 587-594.
- [2] W. G. Egan, T. Hilgeman, The Rings of Saturn: A Frost-Coated Semiconductor?, *Icarus* 30(2) 1977, pp. 413-421.
- [3] B. Begemann, J. Dorschner, T. Henning, H. Mutschke, E. Thamm, A Laboratory Approach to the Interstellar Sulfide Dust Problem, *E. 1994, ApJ*, 423, L71-L74.
- [4] H. Mutschke, B. Begemann, J. Dorschner, H. Henning, *Infrared Physics & Technology* 35(2){3} 1994, pp. 361-374.
- [5] J. B. Pollack, D. Hollenbach, S. Beckwith, D. P. Simonelli, T. Roush, W. Fong, Composition and Radiative Properties of Grains in Molecular Clouds and Accretion Disks, *The Astrophysical Journal* 421 1994, pp. 615-639.
- [6] T. Henning, H. Mutschke, Low-temperature infrared properties of cosmic dust analogues., *A&A*, 1997, pp. 327-743.

## Samples

	Troilite	Synthetic [3,4,6]	DAL [7]	NYS [8]	TYS [8]	DEV [9]
<b>Iron(II) Sulfide Type</b>	Troilite	Troilite with iron	4C-Pyrrhotite	NC-Pyrrhotite: 4.78 – 4.96	NC-Pyrrhotite: 5.12 – 5.52	6C-Pyrrhotite
<b>Iron/Sulfur Ratio</b>	1	1	0.877(1)	0.896(2)	0.903(7)	0.929(2)
<b>Crystal System</b>	Hexagonal	Hexagonal	Monoclinic	Hexagonal		Monoclinic
<b>Origin</b>	MPIK Heidelberg	Begemann, MPS Jena	Harries, IGW Jena			de Velliers
<b>Images</b>						

## Experimental Setup

- Four spectrometers from **VUV to mm** wavelength range at **300 K**
- Two **frequency domain** spectrometers:
  - VUV: 115 nm – 230 nm (under vacuum)
  - UV-VIS-NIR: 190 nm – 2.5 μm (under ambient air)
- Two **time domain** spectrometers:
  - FTIR: 2.0 μm – 500 μm (under vacuum)
  - THz: 250 μm – 2 mm (under dry air)
- Low temperature** measurements down to **10 K** for MIR & FIR

## Multi-Wavelengths Reflectivity

- Binary spectroscopic behavior:**
  - Semiconductor-like** for stoichiometric troilite
  - Metal-like** for iron deficient pyrrhotites
- Mid-IR reflectivity decreases with increasing Fe/S ratio (@ 10 μm)

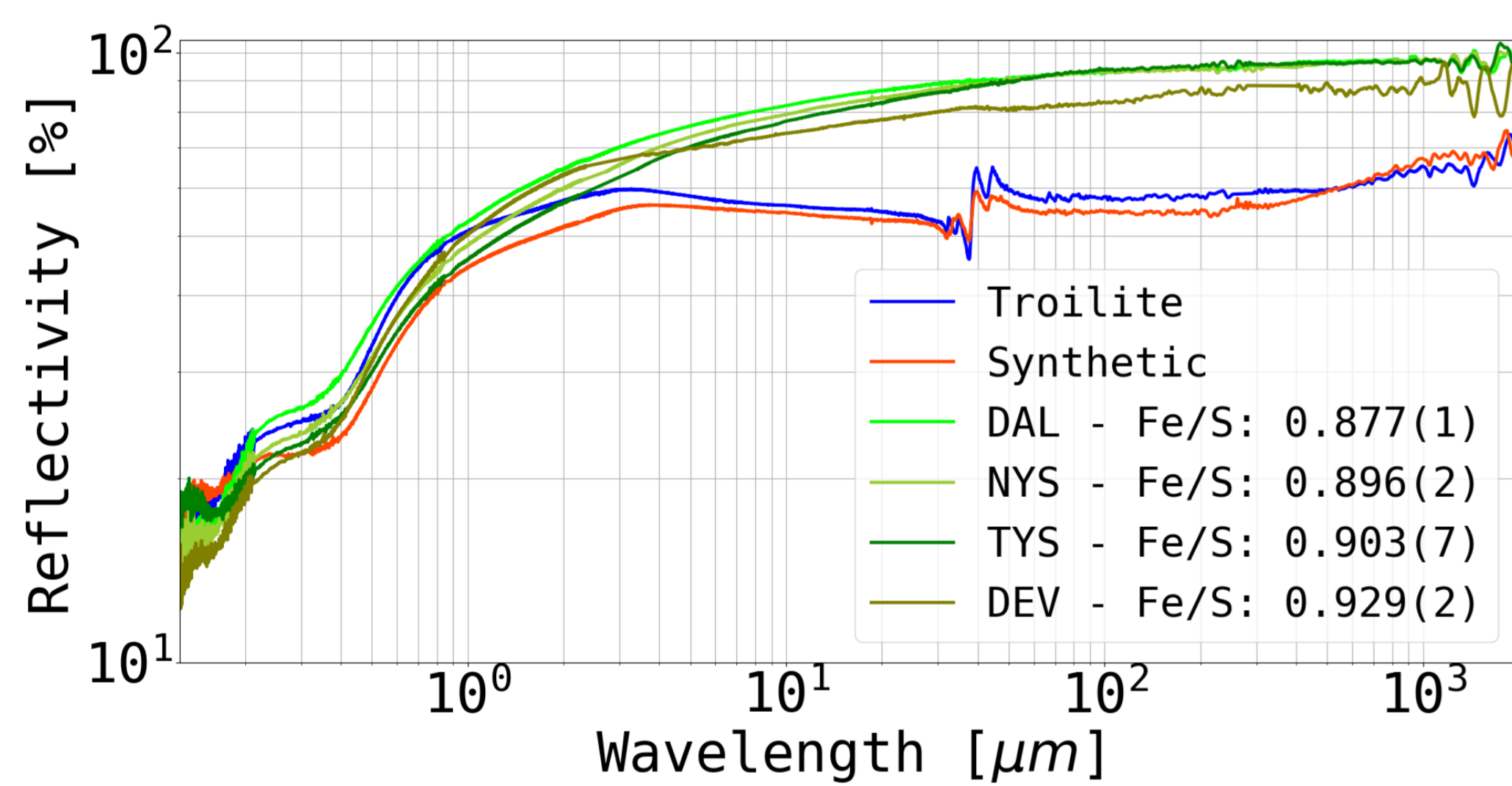


Figure 1: Room temperature reflectivity for samples troilite and pyrrhotites from VUV to mm wavelength regime.

## Optical Constants

- Fitting routine** based on **Levenberg-Marquardt** method
- Optimize n, k parameters based on comparison between measured and computed reflectivity
- Drude-Lorentz-oscillator-model** for troilite & pyrrhotites
- New data in agreement with measured literature values

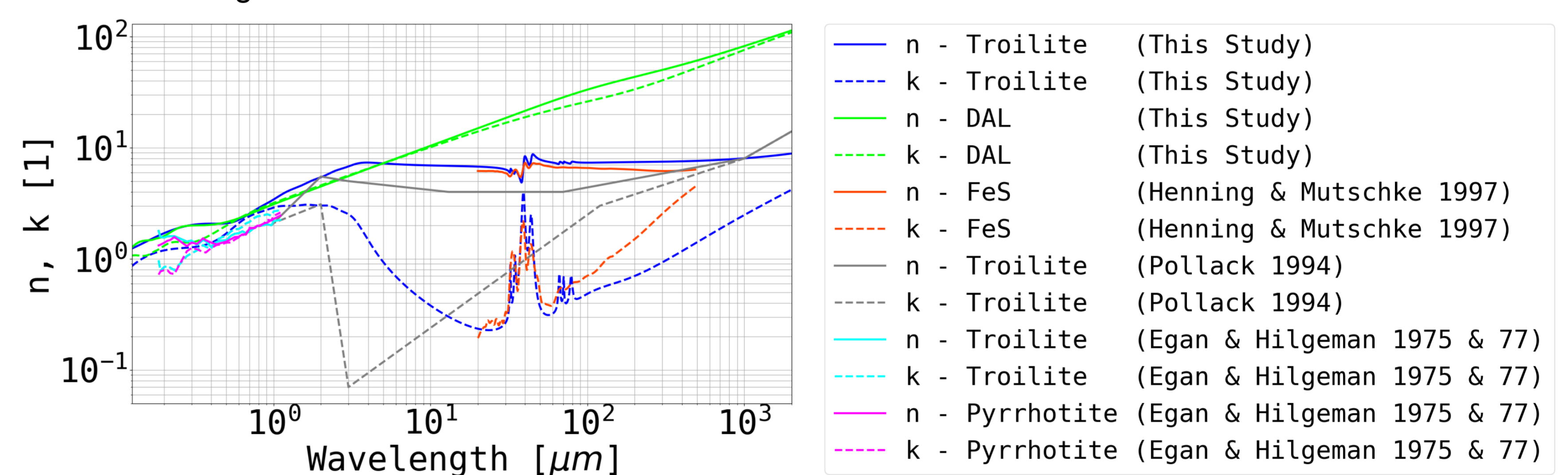


Figure 2: Comparison of n & k values for samples troilite and 4C-pyrrhotite with literature [1, 2, 5, 6].

## Low Temperature Measurements

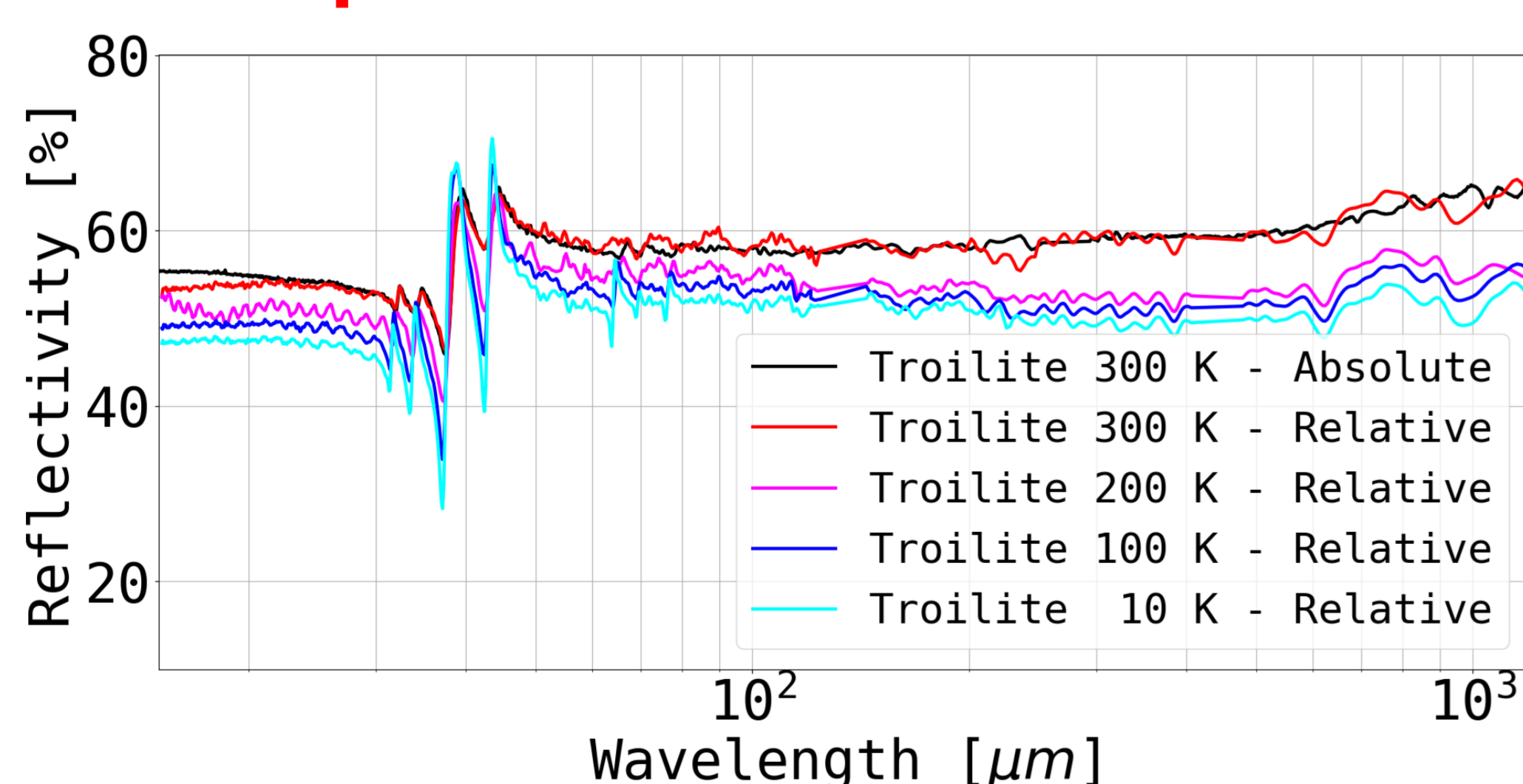


Figure 3: MIR and FIR reflectivity of troilite at low temperatures.

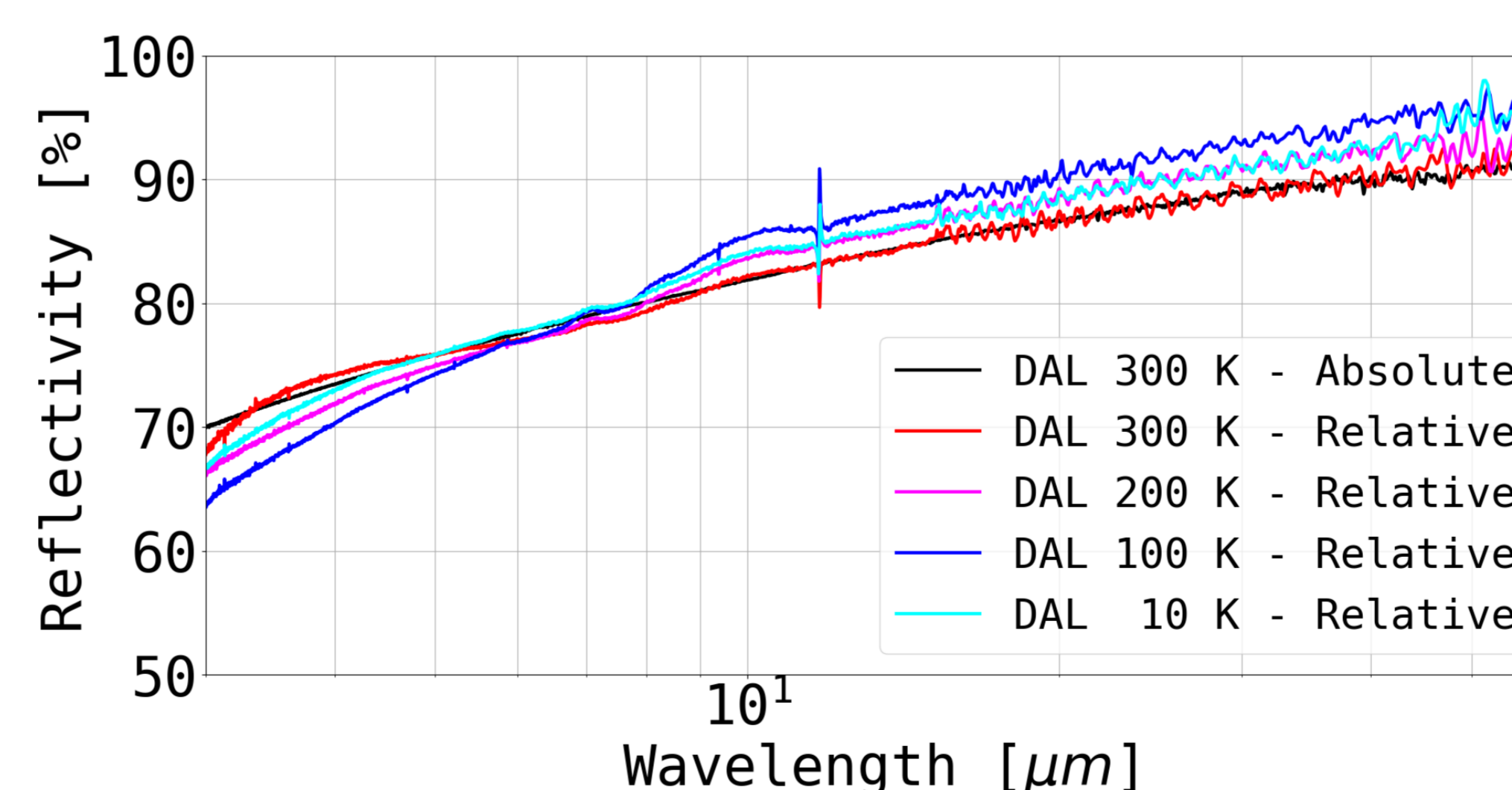


Figure 4: MIR reflectivity of 4C-pyrrhotite at low temperatures.

- Troilite – Semiconductor:**
  - Seven phonon bands:** 32.4 μm, 34.9 μm, 39.7 μm, 44.5 μm, 67.1 μm, 71.1 μm, 79.7 μm (at room temperature)
  - Blue-shift & sharpening of phonon bands
- 4C-Pyrrhotite – Metal:**
  - Free charge carrier** characteristics
  - Change of sign for reflectivity change in IR
  - Possible origin: **Besnus transition at 32 K** [10]

## Acknowledgements

This work was conducted in the Research Unit FOR 2285 “Debris Disks in Planetary Systems” of the Deutsche Forschungsgemeinschaft (grant MU1164/9-2).

## References

- [7] D. Harries, K. Pollok, F. Langenhorst, Oxidative dissolution of 4C- and NC-pyrrhotite: Intrinsic reactivity differences, pH dependence, and the effect of anisotropy. *Geochimica et Cosmochimica Acta* 102, 2013, pp. 23-44.
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